

Contamination of Water Part B: Natural/geogenic chemical pollutants

The material presented here has been prepared by Samuel Addison in April 2021, with input from Dr. Laura Richards and Prof. David Polya of the Department of Earth & Environmental Sciences, The University of Manchester, and other sources as acknowledged. The associated video recordings have been made by Samuel Addison.

The Transformation by Innovation in Distance Education (TIDE) project is enhancing distance learning in Myanmar by building the capacity of Higher Education staff and students, enhancing programmes of study, and strengthening systems that support Higher Educational Institutions in Myanmar. TIDE is part of the UK-Aid-funded Strategic Partnerships for Higher Education Innovation and Reform (SPHEIR) programme (<u>www.spheir.org.uk</u>). SPHEIR is managed on behalf of FCDO by a consortium led by the British Council that includes PwC and Universities UK International. The TIDE project will close in May 2021.



Topic/Lesson



• Outline

- Introduction
- Objectives
- Naturally occurring chemicals
- Learning exercise
- References & Further Information
- Summary



This lesson will provide an overview of key natural/geogenic contaminants.

This lesson will build on the knowledge learnt in lesson "Contamination of Water - Part A: What is contamination?".



• Describe how contaminants can naturally contaminate drinking water

• To be able to discuss selected geogenic pollutants that can naturally occur in water supplies.

NATURALLY OCCURRING CHEMICALS

Naturally occurring chemicals TDE MANCHESTER 1824

 Naturally occurring chemicals are of particular concern since the area of contamination can be quite extensive, and because contamination can go unnoticed in the absence of a testing program [1].

• WHO has established guideline values for compounds that can occur naturally in water [1].



- The WHO provides information for each of these naturally occurring chemicals using chemical fact sheets (see [1]).
- Providing information such as guideline values and the chemical's occurrence.
- The following slides will highlight several of these chemicals prioritized by WHO – but importantly there are many other contaminants.

[1] Chapter 12 of World Health Organization, 2011. *Guidelines for drinking-water quality*. Fourth edition. World Health Organization.

Arsenic (As) GV 0.01 mg/L

- Arsenic in drinking water is a global threat to health, considered by some researchers to have more serious health repercussions than any other environmental contaminant [1]
- Arsenic occurs naturally in soils and rocks and in certain conditions can mobilise and contaminate water [1].
- Global prediction maps attempt to predict where arsenic contamination may be [2]
- See lesson "Arsenic pollution" for more information.

Fluoride (F) GV 1.5 mg/L



- Fluoride, along with arsenic, is one of the most serious chemical contaminants that occurs naturally in drinking water [1]
- Fluoride is a common element in the earth's crust and enters water through interaction with rocks [1]
- Studies have also attempted to model global distribution of fluoride contamination in groundwater [2].



• Barium occurs naturally in rock, with an average of 250 mg/kg in continental crust [1]

 It is positively charged in water and typically occurs at less than 0.1 mg/L, though natural concentrations in groundwater can exceed 1 mg/L [1]

Boron (B) GV 0.5 mg/L



- Boron concentration in rocks averages 10 mg/kg, with up to 100 mg/kg in some rock types [1]
- Boron levels in natural waters range widely, and are dependent on local geology and geochemical conditions [1]
- Boron in surface water is highly variable, though concentrations above 1 mg/L are rare. Groundwater levels range more widely, from < 0.3 to over 100 mg/L [1]

Manganese (Mn) 0.4 mg/L 1

- Manganese is one of the most abundant metals in the earth's crust [1]
- Surface water generally contains low levels of manganese (< 0.1 mg/L), groundwater can contain much higher levels (above 1 mg/L) [1]
- Unlike contaminants such as arsenic and fluoride, high levels of manganese lead to unpleasant tastes or staining, which sometimes can cause people to avoid water with high concentrations of manganese[1]

Selenium (Se) GV 0.01 mg/L TDE MANCHESTER 1824 The University of Manchester

• Selenium is a trace element in rocks, with an average concentration of less than 1 mg/kg [1].

 Sedimentary rocks may contain up to 100 mg/kg, while levels up to several thousand mg/kg have been reported in some coal deposits [1]

• Natural levels of selenium in drinking water are generally below 0.01 mg/L [1]

Uranium (U) GV 0.015 mg/L TDE MANCHESTER

 Uranium occurs naturally in rocks and sediments, with average concentration in soils and rocks of 3 mg/kg [1]

 Although the decay of uranium isotopes produces radioactivity, the main public health threat of uranium arises from its chemical toxicity as a heavy metal [1]

Contaminant fate and transport

MANCHESTER 1824 The University of Manchester

- Natural pathways such as hydrogeological flow paths for groundwaters as well as surface flow paths (e.g. river flow paths) and groundwatersurface water interaction can lead to the transport of contaminants in the environment [1]
- Human activity such as groundwater pumping can also impact the transport of contaminants in the environment [2]

SUMMARY





• There are a wide range of chemicals that contaminate water resources.

• The chemicals can significantly vary in toxicity and concentration in water.

 Arsenic and fluoride are generally regarded as the most serious and widespread natural chemical contaminants especially in groundwater

LEARNING EXERCISE

Learning exercises



- A trait of some natural contaminants is that they spread over large areas, so contaminant mapping is a useful tool.
- An example of this, and for you to investigate is: <u>https://www.gapmaps.org/Home/Public#</u>

- Two studies that have done this also are:
- (For arsenic) <u>https://doi.org/10.1126/science.aba1510</u>
- (For fluoride) <u>https://doi.org/10.1021/es071958y</u>

REFERENCES & FURTHER RESOURCES

References



- UNICEF, 2008. UNICEF Handbook on water quality. United Nations Childrens Fund, New York/USA. <u>https://www.unicef.org/documents/2008-unicef-handbook-waterquality</u>
- World Health Organization, 2011. Guidelines for drinking-water quality. Fourth edition. World Health Organization. Here is a link to download the document: https://www.who.int/water_sanitation_health/publications/2011/dwg_guidelines/e_n
- Podgorski, J. and Berg, M., 2020. Global threat of arsenic in groundwater. *Science*, *368*(6493), pp.845-850. <u>https://doi.org/10.1126/science.aba1510</u>
- Amini, M., Mueller, K., Abbaspour, K.C., Rosenberg, T., Afyuni, M., Møller, K.N., Sarr, M. and Johnson, C.A., 2008. Statistical modeling of global geogenic fluoride contamination in groundwaters. *Environmental science & technology*, 42(10), pp.3662-3668. <u>https://doi.org/10.1021/es071958y</u>



Look at chapter 12 of World Health Organization, 2011. *Guidelines for drinkingwater quality*. Fourth edition. World Health Organization.

Here is a link to the web page with access to download the document: https://www.who.int/water_sanitation_he alth/publications/2011/dwq_guidelines/en/

Disclaimer & Conditions of Use TDE MANCHESTER 1824 The University of Manchester

The mention of specific companies or of certain manufacturers' products does not imply that they are endorsed or recommended by the authors in preference to others of a similar nature that are not mentioned. All reasonable precautions have been taken by the authors to verify the information contained in this work, however, the material is being distributed without warranty of any kind, either express or implied. The responsibility for the interpretation and use of the material lies with the reader. In no event shall the authors be liable for damages arising from the use of the material in this work. The views expressed by the authors do not necessarily represent the views, decisions or the stated policies of any organization or individual referred to in this work.

This work with the exception of material from other sources as indicated , copyrighted material of which is reproduced here as fair dealing for the purposes of research or private study, or criticism or review, as permitted under the UK Copyright, Designs and Patents Act (1998), is provided under the terms of the CC-BY-NC-ND Licence as detailed at: https://creativecommons.org/licenses/by-nc-nd/4.0/legalcode and which, in particular, subject to the terms and conditions of this Public License, grants a worldwide, royalty-free, non-sublicensable, non-exclusive, irrevocable license to exercise the Licensed Rights in the Licensed Material to:

- (i) reproduce and Share the Licensed Material, in whole or in part, for NonCommercial purposes only; and
- (ii) produce and reproduce, but not Share, Adapted Material for NonCommercial purposes only

If You Share the Licensed Material, You must: (A) retain the following if it is supplied by the Licensor with the Licensed Material: (i) identification of the creator(s) of the Licensed Material and any others designated to receive attribution, in any reasonable manner requested by the Licensor (including by pseudonym if designated); (ii) a copyright notice; (III) a notice that refers to this Public License; (iv) a notice that refers to the disclaimer of warranties; (v) a URI or hyperlink to the Licensed Material to the extent reasonably practicable; (B) indicate if You modified the Licensed Material and retain an indication of any previous modifications; and (C) indicate the Licensed Material is licensed under this Public License, and include the text of, or the URI or hyperlink to, this Public License.

For the avoidance of doubt, permission is not granted under this Public License to Share Adapted Material.

Enquiries concerning reproduction outside the terms stated here should be sent to the author at the contact details provided on the title page.

© The Author except for material from other sources as indicated

