

ARSENIC POLLUTION PART B: GEOCHEMICAL CONTROLS ON DISTRIBUTION AND MOBILISATION MECHANISMS

The material presented here has been prepared by George Wilson in April 2021, with input from Dr. Laura Richards and Prof. David Polya of the Department of Earth & Environmental Sciences, The University of Manchester, and other sources as acknowledged. The associated video recordings have been made by George Wilson.

The Transformation by Innovation in Distance Education (TIDE) project is enhancing distance learning in Myanmar by building the capacity of Higher Education staff and students, enhancing programmes of study, and strengthening systems that support Higher Educational Institutions in Myanmar. TIDE is part of the UK-Aid-funded Strategic Partnerships for Higher Education Innovation and Reform (SPHEIR) programme (<u>www.spheir.org.uk</u>). SPHEIR is managed on behalf of FCDO by a consortium led by the British Council that includes PwC and Universities UK International. The TIDE project will close in May 2021.



Geochemistry of arsenic

- Outline
 - Introduction
 - Objectives
 - Geochemistry of As-contaminated groundwater

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- Adsorption and desorption of arsenic on mineral phases
- Organic matter and arsenic
- Learning exercise
- References & Further Information
- Summary

Introduction



- Arsenic geochemical controls are dependent on hydrogeochemical setting and can be highly localized
- Important to understand what causes naturallyoccurring arsenic to be mobilized and accumulate in groundwater at potentially dangerous concentrations
- Bioavailable organic matter recently (past ~ 20 years) postulated to facilitate the microbiallymediated mobilization of geogenic arsenic, although details of mechanistic understanding remain unclear

Objectives



- Explain the predominant geochemical controls on arsenic mobilisation
- Recognize how we can categorize organic matter into different fractions and how this might relate to arsenic mobilization
- Recognize the potential role of organic matter in mobilising arsenic from iron (oxyhyr)oxides

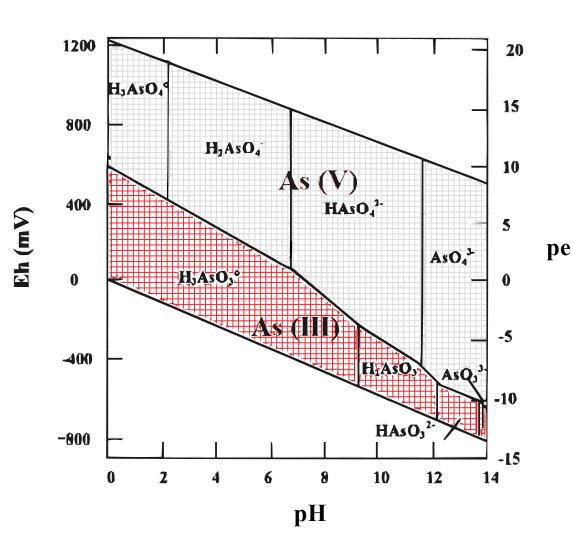
GEOCHEMISTRY OF ARSENIC-CONTAMINATED GROUNDWATER

As speciation



Arsenic exists as two predominant species – As (V) or As (III) depending on Eh/pH

(USGS, 2003)



Geochemical controls on As TDE MANCHESTE 1824 The University of Manchester

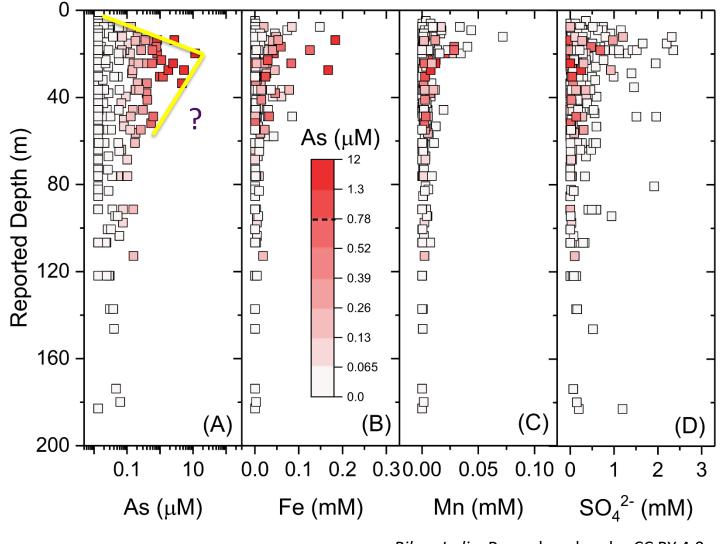
• From last time:

- Reductive dissolution and alkalidesorption (RD) controls the majority of high-As systems worldwide

- RD of iron (oxyhydr)oxides the most important mechanism in South/South East Asia
- Reduction of iron oxides redox controlled
 Do we get higher or lower As with depth?

(Ravenscroft, 2007) (OA)

As mobilisation and depth



Bihar, India. Reproduced under CC BY 4.0 license from Richards *et al.* (2020)

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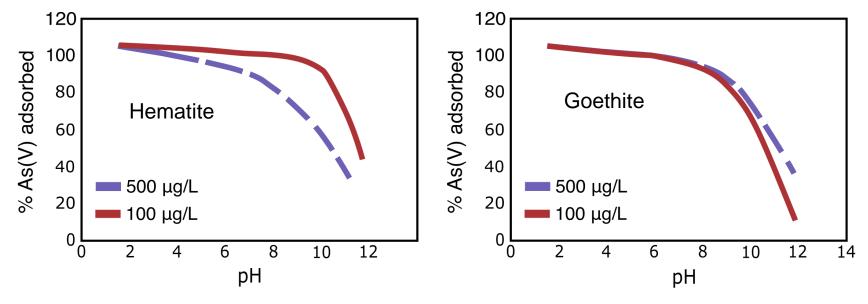
ADSORPTION AND DESORPTION OF ARSENIC ON MINERAL PHASES

As adsorption/desorption on minerals TDE

- Main sorbents are iron oxyhyr(oxides) for arsenic in aquifers and soils
 - Ferrihydrite $Fe^{3+100.14[OH]_2}$
 - Goethite α-FeO[OH]
- Clay minerals follow similar patterns to above, less adsorbed per unit mass
 - Kaolinite Al₂Si₂O₅(OH)₄
 - Illite (K,H₃O)(Al,Mg,Fe)₂(Si,Al)₄O₁₀[(OH)₂,(H₂O)]
 - Montmorillonite $(Na,Ca)_{0.33}(AI,Mg)_2(Si_4O_{10})(OH)_2 \cdot nH_2O$
 - Phosphate sorption?

(Feng et al., 2013) (OA)

As adsorption/desorption on minerals TDE MANCHESTER The University of Manchester



Produced by George Wilson from Mamindy-Pajany et al., (2011)

• Adsorption of arsenate onto most oxides and clays decreases significantly at pH ≥ 8.5 - 'Alkali-desorption'

ORGANIC MATTER AND ARSENIC

Categorization of organic matter

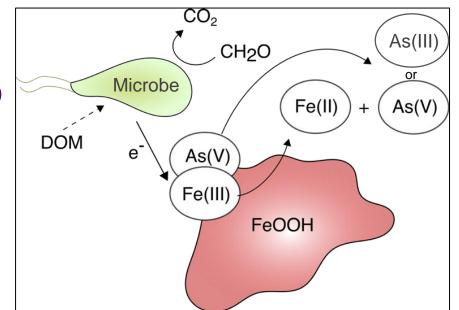
- Types of organic matter (OM):
 - Sediment-bound OM
 - Sources of petroleum
 - Surface-derived OM (ingress through ponds/wetlands/groundwater pumping)
 - Plant-derived OM
 - Organic matter contribution to As mobilisation?

(Lawson et al., 2013) (OA)

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Organic matter mediated mobilisation TDE

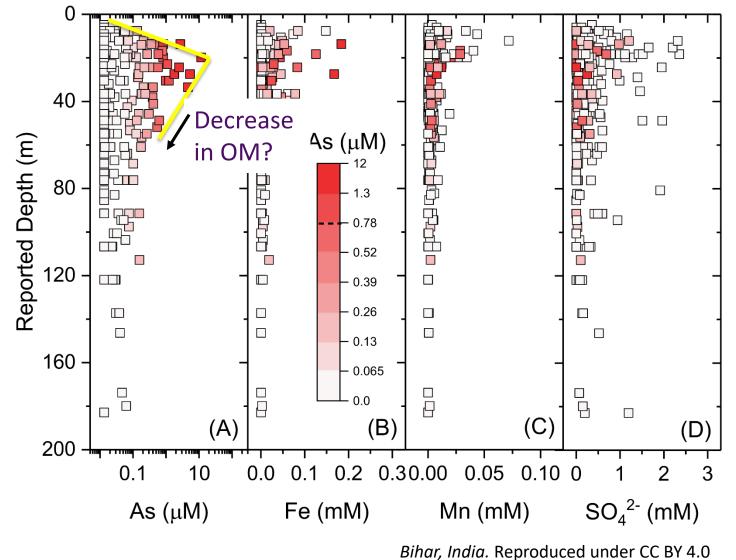
- Reductive dissolution mediated by organic matter
 - OM acts as an electron donor to reduce oxides
 - OM and As in competition for sorption sites on Fe-oxides



Produced by George Wilson from an adaption of Oremland and Stolz, (2003)

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As mobilisation and depth



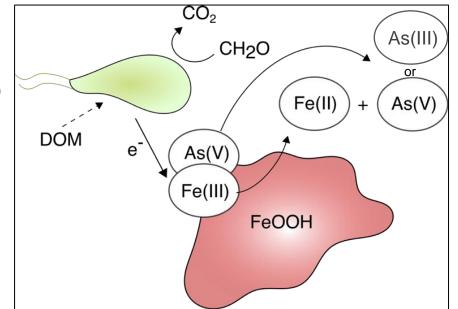
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Organic matter mediated mobilisation

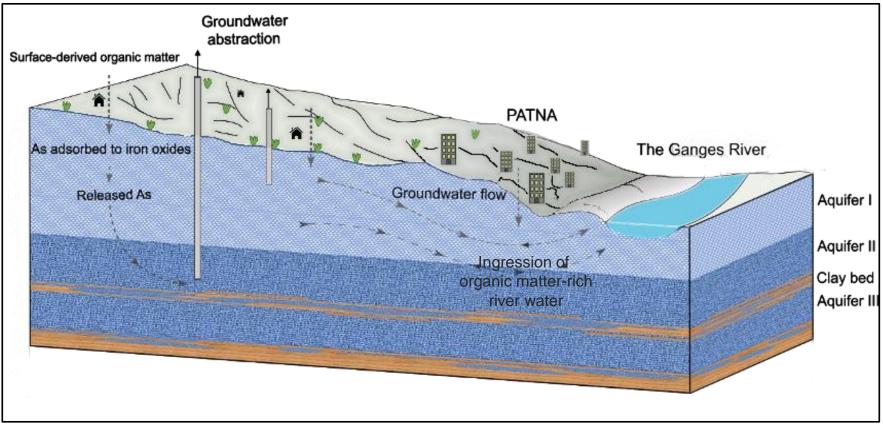
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Produced by George Wilson from an adaption of Oremland and Stolz, (2003)

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Surface-derived organic matter hypothesis TDE



Produced by George Wilson

SUMMARY

Summary



- Arsenic speciation is important to consider - arsenite/As(III) is typically more mobile and difficult to remove than arsenate/As(V)
- Iron (oxyhydr)oxides regulate the release of arsenic in most of As-affected groundwaters
- Organic matter, particularly its type, is thought to play a significant role in the mobilisation of groundwater arsenic

LEARNING EXERCISE



 List and describe two expected geochemical controls on arsenic mobilization for a setting along the Ganges

2a. Explain how organic matter could potentially increase arsenic concentration in groundwater2b. How might the type of organic matter affect this?

REFERENCES & FURTHER RESOURCES

References



Feng, Q., Zhang, Z., Chen, Y., Liu, L., Zhang, Z. and Chen, C., 2013. Adsorption and desorption characteristics of arsenic on soils: kinetics, equilibrium, and effect of Fe (OH) 3 colloid, H2SiO3 colloid and phosphate. *Procedia Environmental Sciences*, *18*, pp.26-36. <u>https://doi.org/10.1016/j.proenv.2013.04.005</u> (OA)

Lawson, M., Polya, D.A., Boyce, A.J., Bryant, C., Mondal, D., Shantz, A. and Ballentine, C.J., 2013. Pondderived organic carbon driving changes in arsenic hazard found in Asian groundwaters. *Environmental science & technology*, *47*(13), pp.7085-7094 <u>https://pubs.acs.org/doi/10.1021/es400114q</u>

Mamindy-Pajany, Y., Hurel, C., Marmier, N. and Roméo, M., 2011. Arsenic (V) adsorption from aqueous solution onto goethite, hematite, magnetite and zero-valent iron: effects of pH, concentration and reversibility. *Desalination*, *281*, pp.93-99. <u>https://doi.org/10.1016/j.desal.2011.07.046</u>

Oremland, R.S. and Stolz, J.F., 2003. The ecology of arsenic. *Science*, 300(5621), pp.939-944.

Ravenscroft, 2007. Predicting the Global Distribution of Natural Arsenic Pollution. [Online] Available at: <u>https://www.geog.cam.ac.uk/research/projects/arsenic/symposium/S1.2_P_Ravenscroft.pdf</u> [Accessed 07/04/21] (OA)

Richards, L.A., Kumar, A., Shankar, P., Gaurav, A., Ghosh, A. and Polya, D.A., 2020. Distribution and geochemical controls of arsenic and uranium in groundwater-derived drinking water in Bihar, India. *International journal of environmental research and public health*, *17*(7), p.2500. <u>https://doi.org/10.3390/ijerph17072500</u> (OA)

USGS, 2003. Water Resources Center Annual Technical Report FY 2003. *United States Geological Survey*. Available at: <u>https://water.usgs.gov/wrri/AnnualReports/2003/FY2003_MD_Annual_Report.pdf</u> [Accessed 04/05/21] (OA)

Further Resources

Arsenic pollution guide

Bhattacharya, P., Polya, D. and Jovanovic, D. eds., 2017. *Best practice guide on the control of arsenic in drinking water*. IWA Publishing. <u>https://doi.org/10.2166/9781780404929</u> (OA chapters)

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